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Modern Chemistry Chapter 15 Review

Modern Chemistry Chapter 15. Acid-Base Titration & pH. Section 1 self-ionization of water occurs when two water molecules produce a hydronium (H_3O^+) and a hydroxide (OH^-) ion $2 \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ The ionization constant (K_w) of water is: $K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14} \text{ M}$ Acidic, Basic, & Neutral
IF $[\text{H}_3\text{O}^+] > [\text{OH}^-]$ then solution is acidic.
IF $[\text{H}_3\text{O}^+] < [\text{OH}^-]$ then solution is basic.

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pH of an acid or base solutions given either the hydroxide ion concentration or the hydronium ion concentration. Holt McDougal Modern Chemistry Chapter 15: Acid-Base ...

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CHAPTER 15 REVIEW . Acid-Base
Titration and pH. SHORT ANSWER .

Answer the following questions in the
space provided. 1. Calculate the
following values without using a
calculator. 4.0 . a. The $[H_3O^+]$ in a
solution is 1.0×10^{-4} M. Calculate the
pH. _____; 1.0×10^{-13} M. Calculate the
pH. _____; 1.0×10^{-13} M. Calculate
the pH of a solution is 13.0. Calculate
the $[H_3O^+]$ in a solution with a pH of 13.0.

CHAPTER 15 REVIEW Acid-Base Titration and pH

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SECTION 3 continued 4. Table 6on page
485 of the text lists several amphoteric
species, but only one other than water is
neutral. NH₃ a. Identify that neutral
compound. b. Write two equations that
demonstrate this compound's
amphoteric properties. (Answers will
vary, but one equation should form NH₄⁺)

14 Acids and Bases

The Acid-Base Titration and pH chapter
of this Holt McDougal Modern Chemistry
Companion Course helps students learn
the essential lessons associated with pH

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and acid-base titration.

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15 Acid-Base Titration and pH CHAPTER
15 REVIEW Acid-Base Titration and pH
SECTION 2 SHORT ANSWER Answer the
following questions in the space
provided 1 Below is a pH curve from an
acid-base titration On it are labeled
three points: X, Y, and Z

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Answers

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30 Temperature CC) Do the data points
lie on a straight line? Based on your
sketch, predict the approximate partial
pressure for ... CHAPTER 11 REVIEW
Gases Class SHORT ANSWER Answer the
following questions in the space
provided. 1. State whether the pressure
of a fixed mass of gas will increase,
decrease, or ...

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73 ... CHAPTER 9 REVIEW Stoichiometry
SECTION 1 SHORT ANSWER Answer the
following questions in the space
provided. 1. b The coefficients in a
chemical equation represent the (a)
masses in grams of all reactants and
products. ... 15.0 mol c. If 20.0 mol of
NO₂ form, how many moles of O₂ must
have been consumed?

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